

10 mL

$$\text{@ } 10.1 \text{ mL} - 10 \text{ mL} = 0.1 \text{ mL}$$

$$1 \text{ M} \times \frac{0.1 \text{ mL}}{110.1 \text{ mL}} = 9.08 \times 10^{-4} \text{ M}$$

$$\text{pH} = -\log(9.08 \times 10^{-4})$$

$$\text{pH} = 3.04$$